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Full Length Research

# Antioxidants and Ameliorative Potentials of Stem Bark Extract of *Anogeissus leiocarpus* against Formalin - Induced Pain and Inflammation in Wistar Rats

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The study was carried out to reveal the toxicity dosage, antioxidant and anti-inflammatory activities of hexane and ethanolic stem bark extract of Anogeissus leiocarpus (AI) in albino rats. The antioxidant ability of hexane and ethanolic stem bark extract of Anogeissus leiocarpus was carried out using various antioxidant assays. The toxicity study was also conducted using 50% inhibitory concentration (IC<sub>50</sub>). Formalin-induced paw oedema model was used to investigate antiinflammatory effects of hexane and ethanolic stem bark extract of Anogeissus leiocarpus in experimental rats. There is a dearth of information on the anti-inflammatory activity of this study plant in an oedema model induced by formalin and inflammatory mediators in rats. Therefore, this research was to evaluate the toxicity dosage, antioxidant and anti-inflammatory activities of hexane and ethanolic stem bark extracts of Anogeissus leiocarpus (AI) in wistar albino rats to validate scientifically the folklore claims of the plant. The results showed that free radicals were statistically scavenged at p < 0.05 by stem bark extract of the study plant at different concentrations. The higher percentage hydroxyl scavenging ability was recorded in hexane extract of stem bark of Al at 100 µg/ml (80.70 ± 0.22) while lower percentage hydroxyl scavenging ability was recorded by ethanolic extract of AI (61.19 ± 0.62). Hexane extract showed higher percentage of inhibition of DPPH activity (80.95  $\pm$  0.86) while ethanolic extract recorded lower percentage potential of 60.26  $\pm$ 0.18 at 100 µg/ml, in a concentration dependent manner. FRAP values of ethanolic stem bark extract (46.67  $\pm$  0.33 (mg AA/g extract)) was lower than the hexane stem bark extract (80.55  $\pm$ 1.97 (mg AA/ g extract) of Anogeissus leiocarpus). Daily oral administration of hexane and ethanolic stem bark extract of Anogeissus leiocarpus was done for 14 days. No mortality was recorded after fourteen (14) days of treatment. The studied plant is safe and nontoxic to rats (LD<sub>50</sub>> 5000 mg/kg). The highest percentage inhibition was exhibited by hexane extract at dose of 300 mg/kg body weight while the lowest percentage paw oedema inhibition (33.57 %) was observed in ethanolic extract at dose of 75 mg/kg. The standard drug (ibuprofen) at 400 mg/kg body weight exhibited the percentage of inhibition of 64.70%. The antioxidant potentials and the ability of the stem bark extract of Anogeissus leiocarpus to reduce the size of oedema produced by formaldehyde, suggested that it has chemical constituents that may be active against inflammatory disorders.

Key words: Anogeissus leiocarpus, antioxidant, formalin, anti-inflammatory.

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#### INTRODUCTION

The induction of pain by formalin is due to the inflammation of the peripheral tissue, leading to a central sensitization of dorsal horn neurons during the inflammatory pain (Ji *et al.*, 2018). Formalininduced pain is characterized by exudation of fluid and plasma proteins and the emigration of leukocytes, majorly neutrophils (Jang *et al.*, 2020). Several works have reported that formalin injection also produces an oedema and an increase of vascular permeability in the injected paw. The first response of the immune system to infection or irritation that helps the body to protect itself against an injurious stimulus is inflammation (Jang *et al.*, 2020). Inflammation can be brought about by physical injury, antibodies or infections (Chen *et al.*, 2017). There are two types of inflammation (acute or chronic) depending on the nature of the effectiveness of the initial reaction or the stimulus in eliminating the stimulus or the damaged tissues (Ashmawi *et al.*, 2016; Asahara *et al.*, 2015).

Nonsteroidal Anti-inflammatory drugs are currently used for management of pain and inflammatory disorders. These drugs have adverse side effects including gastrointestinal bleeding and peptic ulcers. Therefore, there is need to explore other sources especially of plant origin for new production of anti-inflammatory drugs that are readily available, more effective and less toxic than the synthetic drugs. African birch (*Anogeissus leiocarpus* DC. Guill & Perr.) belongs to Combretaceae, includes eight species that are distributed in Asia and Africa. *Anogeissus leiocarpus* has also been reported as a good source of traditional medicine for wound healing, skin diseases and microbial infection (Dzeufiet *et al.*, 2014).

Therefore, this study seeks to carry out the antioxidant and anti-inflammatory potentials of stem bark extract of *Anogeissus leoicarpus* against formalin - induced pain and inflammation in wistar rats, to validate the folklore use of this study plant on experimental inflammation in rats.

# MATERIALS AND METHODS

#### **Plant Collection and Extraction**

One (1) kg of stem bark of *Anogeissus leiocarpus* was purchased from herbal practitioners at Akure main market, Akure, Ondo State, Nigeria. Identification and authentication of the study plant was carried out by the Crop, Soil and Pest Management (CSP) Department at the Federal University of Technology, Akure (FUTA), Nigeria. Voucher sample was deposited at the institution herbarium with reference number FUTA/HB/0259 allotted. Also, the voucher specimen was deposited in the Department of Neurofarba, Pharmaceutical and Neuroceutical section, Research Unit of Phytolab, University of Florence, Italy. The air-dried stem bark was coarsely powdered and subjected to cold extraction using hexane and 70% ethanol solvents separately for 72 hours, the filtrate was concentrated using rotary evaporator and freeze dried to obtain crude extract.

#### **Drugs and Chemicals**

Ibuprofen (Ranbaxy), formalin and other chemicals were of analytical grade.

#### Evaluation of Antioxidant Components and Activity

#### DPPH (1, 1-diphenyl-2-picryl-hydrazyl) radical scavenging assay

The ability of *Anogeissus leiocarpus* (AI) to scavenge the DPPH radicals was assessed by using the method described by Molyneux (2004) with a slight modification. DPPH solution (0.1 mmol/l) was prepared in methanol and 0.5 ml was added to different concentrations of AI (25, 50 75 and 100  $\mu$ g/ml). The mixture was incubated at room temperature for 30 min. A control reference was carried out which contains methanol as a baseline correction instead of AI. After incubation, absorbance was taken at 517 nm. The result was compared with the ascorbic acid, a standard antioxidant. Percentage DPPH radical scavenging activity was calculated and plotted against concentration. IC<sub>50</sub> value (the microgram of extract to scavenge 50% of the radicals) was evaluated using linear regression analysis.

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% DPPH inhibition =  $\frac{(Abs of Control - Ab of Sample)}{(Abs of Control)} \times 100$ 

Absorbance of control (without sample) = Abs control; Absorbance of Al= Abs sample.

# ABTS radical cation decolorization assay

Free radical scavenging ability of the AI extract was determined by ABTS radical cation decolorization assay and was carried out according to the method of Re et al. (1999). ABTS:+ cation radical was produced by reacting 7 mM of ABTS and 2.45 mM potassium persulfate (1:1) and stored in the dark at room temperature for 12-16 h before use. ABTS:+ solution was then diluted with 80% methanol to obtain absorbance between 0.700 at 734 nm. Then, 100  $\mu$ l of different concentration of AI or Trolox (25, 50, 75 and 100 $\mu$ g/ml) was added to 2 ml of diluted ABTS:+ solution. The mixture was incubated for 10 min in the dark and absorbance was taken at 734 nm. A control and solvent blank were run for the assay. The entire assay was done in triplicate. Trolox was used as reference standard. IC<sub>50</sub> value (the microgram of extract to scavenge 50% of the radicals) was evaluated using linear regression analysis

Percentage ABTS cation radical inhibition was calculated from the formula,

% ABTS cation radical inhibition =  $\frac{Abs of control - Abs of sample}{Abs of control} \times 100$ 

Absorbance of control= Abs control, Absorbance of AI = Abs sample.

#### Hydroxyl Radical Scavenging Activity

This assay was described by Jin *et al.*, (1996).The hydroxyl radical was generated in a mixture of 1.0 ml of 0.75 mM 1,10-phenanthroline, 2.0 ml of 0.2 M sodium phosphate buffer (pH 7.4), 1.0 ml of 0.75 mM FeSO<sub>4</sub> and 1.0 ml of  $H_2O_2$  (0.01%, v/v). After addition of 1.0 ml extract at various concentrations (25, 50, 75 and 100 µg/ml), the mixture was incubated at 37°C for 30 min. Then, the absorbance of the mixture at 536 nm was measured. Ascorbic acid also at different concentrations (25, 50, 75 and 100 µg/ml) was used as reference standard and was treated the same way as that of extracts. Distilled water was used as the blank. The scavenging activity on - OH was calculated by the following equation:

Scavenging activity (%) = {(Abs sample – Abs blank)/ (Abs0 – Abs blank)} \*100 Where, Abs 0 is the absorbance of the distilled water instead of  $H_2O_2$  and extract in the assay system

# Nitric Oxide Radical Scavenging Activity

This assayed was described by Sangameswaran *et al.*, (2009). Various concentrations (25, 50, 75 and 100 µg/ml) of the extract and standard were prepared. Sodium nitroprusside (2.5 ml, 10 mM) in phosphate buffered saline (PBS) was added to 0.5ml different concentrations of extract and standard. The reaction mixture was incubated at 25°C for 150 min. After incubation, 0.5 ml aliquot was removed and 0.5 ml of Griess reagent: (1% (w/v) added. The absorbance was measured at 546 nm. Ascorbic acid was used as reference standard and was treated the same way as that of extracts. Sodium nitroprusside in PBS (2 ml) was used as control. The nitric oxide radicals scavenging activity of the extracts and ascorbic acid was calculated according to the following equation:

Percentage of inhibition =  $[(Ao - A1) / Ao] \times 100$ .

Where Ao is the absorbance of sodium nitropruside in PBS (without extract or ascorbic acid) and A1 is the absorbance in the presence of the extract or ascorbic acid.

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# Ferric Reducing-Antioxidant Property (FRAP) Assay

The reducing property of the sample was determined by method described by Pulido et al. (2000).FRAP reagent solution containing 300 mM acetate buffer, 10 mM TPTZ in 40 mMHCl, 20 mM FeCl<sub>3</sub>.6H<sub>2</sub>O (10:1:1 respectively) was prepared. Three millilitre of prepared FRAP reagent was mixed with 100  $\mu$ l *Anogeissus leiocarpus* (AI) extract (1 mg/mI) and the absorbance at 593 nm was recorded after a 30 min incubation at 37°C. A linear graph of FeSO<sub>4</sub> reference standard was prepared and the FRAP value of AI was extrapolated from it. FRAP value was expressed as mMol Fe<sup>2+</sup> equivalents per g dry weight.

## **Determination of Total Phenolics Content**

The total phenolic content of extracts was determined using the Folin-Ciocalteu method as modified by Hung *et al.* (2001). Extracts (AI) (0.1 ml) was rapidly mixed with 0.1 ml of Folin-Ciocalteu reagent, followed by the addition of 0.3 ml sodium carbonate (15%, w/v) solution. The mixture was incubated in the dark for 30 mins. The absorbance of the blue colour was read at 760 nm after 30 mins on a spectrophotometer. The total phenolic content was extrapolated from a standard curve using gallic acid (graded doses, 25 - 100 ug/ml) as a standard phenolic acid. The amount of total phenolics was expressed as gallic acid equivalents (GAE, mg gallic acid/g AI).

## **Determination of Total Flavonoid Content**

The total flavonoid content was determined spectrophotometrically according to the modified method of Kumaran and Karunakaran (2007). Briefly, 0.5 ml of Al solution and standard (quercetin) at different concentrations ( $25 - 100\mu g/ml$ ) were taken in test tubes dissolved in methanol followed by the addition of 0.1 ml of 10% aluminium chloride solution. 0.1 ml of 1M sodium acetate solution was added to the mixtures in the test tubes. Furthermore, each reaction test tube was then immediately diluted with 2.8 ml of distilled water and mixed to incubate for 30 min at room temperature to complete reaction. The absorbance of pink coloured solution was noted at 415 nm using a spectrophotometer against blank methanol. The total flavonoids content was expressed as mg/g Quercetin Equivalent (mg/g QE).

### Animals

Albino wistar rats weighing approximately (160–250 g body weight) were used. Rats were housed and used at least one week after their arrival in plastic cages with filter tops at the Animal House Lab, Department of Biochemistry, Federal University of Technology Akure, Ondo State, Nigeria. Six rats were housed per cage; animals were fed a standard laboratory diet and tap water ad libitum and kept at under standard environmental conditions (23–25°C, 12-h light/12-h dark cycle) (OECD,2010).

# Acute Toxicity Test

Rats `were` orally administered hexane and ethanolic stem bark extract of *Anogeissus leiocarpus albidum* singly at varying dosage (0-5000 mg/kg b.w.) on day 1 and the animals were observed for 24 hours to detect any sign of toxicity or death as well as delayed toxic symptoms for 7 days (Lorke, 1983).

# Induction of inflammation using rat model of formalin-induced paw oedema (Experimental Protocol)

Anti-inflammatory activity of hexane and ethanolic stem bark of *Anogeissus leiocarpus* was measured in rat model of formalin-induced paw oedema. Albino rats fasted overnight were divided into 9 groups of six animals each, the dosage of the drugs administered to the different groups was as follows.

Group I- control (normal saline) Group II- Formalin induced (0.1ml/kg bw) Group III-V -Formalin + (hexane extract of AI; 75, 150 and 300 mg/kg b.w) for 14 days Group VI-VIII -Formalin + (ethanol) extract of AI.; 75,150 and 300 mg/kg b.w) for 14 days Group IX - Standard drug (Ibuprofen) for 14 days

# Formalin-induced paw oedema, pain and inflammation

Thirty minutes pre oral treatment with extract/drug, following injection of formalin (0.1ml of 10% v/v) into the right hind paw of the tested rats. No injection of formalin into the control group animals. The paw thickness was measured before and after induction of inflammation by using vernier calliper. The increase in paw oedema was measured by vernier calliper according to method described by Taylor *et al.* (2000) and Joseph *et al.* (2005) with some modifications.

The difference in paw thickness after and before induction of inflammation was calculated and presented as mean increase in paw thickness (cm). The ability of hexane and ethanolic stem bark extract of *Anogeissus leoicarpus* and the standard (ibuprofen) used as anti-inflammatory drugs to suppress paw inflammation was expressed as a percentage of inhibition of paw oedema (Taylor *et al.*, 2000; Joseph *et al.*, 2005).

#### Statistical analysis

Results were expressed as mean  $\pm$  standard deviation (SD). Statistical analyses were performed using Microsoft Excel and Graph Pad prism 8 software. Data were analyzed using student t-test and analysis of variance (One-way ANOVA) followed by Turkey's test. Values of *p* < 0.05 were considered as statistically significant.

#### RESULTS

 Table 1. DPPH (%) Scavenging Ability of Hexane and Ethanolic Stem Bark

 Extract of Anogeissus leoicarpus (Al)

| Concentration<br>(µg/ml) | Hexane                    | EtOH                      | Ascorbic acid             |
|--------------------------|---------------------------|---------------------------|---------------------------|
| 25                       | 32.99 ± 0.18 <sup>h</sup> | 21.56 ± 0.37 <sup>1</sup> | $50.48 \pm 0.80^{\circ}$  |
| 50                       | $50.91 \pm 0.18^{t}$      | 37.84 ± 0.31 <sup>g</sup> | $64.42 \pm 2.20^{d}$      |
| 75                       | 60.69 ± 0.49 <sup>e</sup> | $48.83 \pm 0.37^{f}$      | 75.15 ± 1.53 <sup>°</sup> |
| 100                      | 80.95 ± 0.86 <sup>b</sup> | 60.26 ± 0.18 <sup>e</sup> | 91.43 ± 0.18 <sup>ª</sup> |

Values (means of three replicates) followed by different letters are significantly different at p < 0.05

| Table 2. Hydroxyl Radical (%)   | Scavenging  | Ability of | Hexane | and | Ethanolic | Stem |
|---------------------------------|-------------|------------|--------|-----|-----------|------|
| Bark Extract of Anogeissus leoi | carpus (Al) |            |        |     |           |      |

| Conc (µg/ml) | Hexane                       | EtOH                         | (Ascorbicacid)            |
|--------------|------------------------------|------------------------------|---------------------------|
| 25           | 34.29 ± 5.44 <sup>k</sup>    | 21.46 ± 0.87 <sup>1</sup>    | 60.37 ± 1.31 <sup>g</sup> |
| 50           | 62.52 ± 0.44 <sup>f, g</sup> | 44.76 ± 4.38 <sup>j</sup>    | 71.56 ± 3.70 <sup>d</sup> |
| 75           | 70.12 ± 0.22 <sup>d, e</sup> | 53.39 ± 1.24 <sup>h, i</sup> | 85.83 ± 0.87 <sup>b</sup> |
| 100          | $80.70 \pm 0.22^{\circ}$     | 61.19 ± 0.62 <sup>g</sup>    | $94.35 \pm 0.22^{a}$      |

Values (means of three replicates) followed by different letters are significantly different at p < 0.05.

| Conc (µg/mi) | Hexane                       | EtOH                         | (Ascorbic acid)              |
|--------------|------------------------------|------------------------------|------------------------------|
| 25           | 35.68 ± 0.18 <sup>I, m</sup> | 21.08 ± 8.74 <sup>n</sup>    | 63.06 ± 0.18 <sup>e, †</sup> |
| 50           | 53.93 ± 0.38 <sup>i</sup>    | 38.32 ± 0.63 <sup>k, I</sup> | 78.56 ± 1.08 <sup>d</sup>    |
| 75           | $62.70 \pm 0.79^{e, f, g}$   | 42.82 ± 0.63 <sup>j</sup>    | 89.79 ± 0.45 <sup>b</sup>    |
| 100          | 82.64 ± 0.55 <sup>°</sup>    | 58.92 ± 0.18 <sup>h</sup>    | 97.42 ± 0.38 <sup>a</sup>    |

 Table 3. Nitric Oxide (%) Scavenging Ability of Hexane and Ethanolic Stem Bark Extract of Anogeissus leoicarpus (Al)

 Page (up (m))

Values (means of three replicates) followed by different letters are significantly different at p < 0.05

 Table 4. ABTS Assay (%) of Hexane and Ethanolic Stem Bark Extract of Anogeissus leoicarpus (AI)

| Conc (µg/ml) | Hexane                     | EtOH                 | (Trolox)                  |
|--------------|----------------------------|----------------------|---------------------------|
| 25           | $63.20 \pm 0.74^{\dagger}$ | $36.33 \pm 0.58^{j}$ | 71.07 ± 0.28 <sup>d</sup> |
| 50           | 67.23 ± 0.58 <sup>e</sup>  | 41.57 ± 0.56         | $77.99 \pm 0.86^{\circ}$  |
| 75           | $78.84 \pm 0.43^{\circ}$   | $45.88 \pm 0.43^{h}$ | 84.46 ± 0.71 <sup>b</sup> |
| 100          | $85.21 \pm 0.86^{b}$       | $53.37 \pm 0.74^{g}$ | $95.88 \pm 0.32^{a}$      |

Values (means of three replicates) followed by different letters are significantly different at p < 0.05

 Table 5. IC<sub>50</sub> of In vitro Assay of Hexane and Ethanolic Stem Bark Extract of Anogeissus

 leiocarpus (Al)

| <i>In vitro</i> assay (%) IC <sub>50</sub> (µg/ ml) | Hexane       | Ethanol      | Standard      |  |
|---|--------------|--------------|---------------|--|
| DPPH Scavenging Ability                             | 55.17 ± 2.18 | 80.10 ± 6.32 | 55.29 ± 4.86  |  |
| OH Scavenging Ability                               | 47.50 ± 1.33 | 47.94 ± 4.29 | 58.21 ± 3.59  |  |
| NO Scavenging Ability                               | 56.70 ± 0.99 | 65.56 ± 3.92 | 100.13 ± 6.29 |  |
| ABTS Scavenging Ability                             | 48.11 ± 6.84 | 46.23 ± 5.53 | 41.38 ± 6.39  |  |

Values (means of three replicates) followed by different letters are significantly different at p < 0.05

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| Phytochemicals             | Hexane                           | Ethanol                          | _ |
|----------------------------|----------------------------------|----------------------------------|---|
| Total Phenolic Content     | 69.84 ± 0.16 (mg GAE/ g extract) | 53.28 ± 0.15 (mg GAE/ g extract) |   |
| Total Flavonoids Content   | 60.75 ± 0.31 (mg QUE/ g extract) | 50.34 ± 0.42 (mg QUE/ g extract) |   |
| Total Antioxidant Capacity | 70.00 ± 2.15 (mg AAE/g extract)  | 58.70 ± 0.31(mg AAE/g extract)   |   |
| FRAP                       | 80.55 ± 1.97 (mg AA/g extract)   | 46.67 ± 0.33(mg AA/g extract)    |   |

 Table 6.
 Phenolic Content and Antioxidant Activities of Hexane and Ethanolic Stem Bark Extract of Anogeissus leoicarpus

# Key: AA -Ascorbic acid; GAE- Gallic acid; QUE- Quercetin; FRAP- Ferric reducing-antioxidant property

Table 7. Anti-inflammatory Potential of Hexane and Ethanolic Stem bark Extract of Anogeissus leiocarpus against Formalin -Induced Paw Oedema/ Inflammation in Rats

|                             | 0 min               | 30mins              | 1 hr                | 2 hr                | 3 hr                | 4 hr                       | 5 hr                         | 6 hr                     | % Inhibition in<br>paw oedema |
|-----------------------------|---------------------|---------------------|---------------------|---------------------|---------------------|----------------------------|------------------------------|--------------------------|-------------------------------|
| Control                     | 0.35 ± 0.01         | 0.35 ± 0.01         | 0.35 ± 0.01         | 0.35 ± 0.01         | 0.35 ± 0.01         | 0.35 ± 0.01                | 0.35 ± 0.01                  | 0.35 ± 0.01              | 100%                          |
| Induced                     | $0.33 \pm 0.03^{a}$ | $0.60 \pm 0.04^{a}$ | $0.65 \pm 0.03^{a}$ | $0.74 \pm 0.09^{a}$ | $0.89 \pm 0.01^{a}$ | $0.86 \pm 0.02^{a}$        | $0.84 \pm 0.02^{a}$          | $0.81 \pm 0.05^{a}$      | -                             |
| I + Al Hex<br>(75 mg/kg)    | $0.35 \pm 0.04^{a}$ | $0.42 \pm 0.03^{a}$ | $0.49 \pm 0.04^{a}$ | $0.58 \pm 0.04^{a}$ | $0.72 \pm 0.02^{a}$ | $0.62 \pm 0.01^{a}$        | $0.61 \pm 0.00^{a}$          | $0.61 \pm 0.00^{a}$      | 45.45%                        |
| I + Al Hex<br>(150 mg/kg)   | $0.36 \pm 0.01^{a}$ | $0.60 \pm 0.02^{a}$ | $0.62 \pm 0.02^{a}$ | $0.65 \pm 0.03^{a}$ | $0.70 \pm 0.02^{a}$ | $0.61 \pm 0.02^{a}$        | $0.60 \pm 0.03^{a}$          | $0.60 \pm 0.03^{a}$      | 50.70%                        |
| I + Al Hex<br>(300 mg/kg)   | $0.36 \pm 0.02^{a}$ | $0.44 \pm 0.04^{a}$ | $0.48 \pm 0.03^{a}$ | $0.55 \pm 0.03^{a}$ | $0.64 \pm 0.04^{a}$ | $0.58 \pm 0.06^{a}$        | $0.57 \pm 0.06$ <sup>a</sup> | $0.57 \pm 0.06^{a}$      | 56.29%                        |
| I + AI EtOH<br>(75 mg/kg)   | $0.36 \pm 0.02^{a}$ | $0.46 \pm 0.03^{a}$ | $0.53 \pm 0.0^{a}$  | $0.62 \pm 0.04^{a}$ | $0.72 \pm 0.05^{a}$ | $0.69 \pm 0.05^{a}$        | $0.68 \pm 0.04^{a}$          | $0.67 \pm 0.05^{a}$      | 33.57%                        |
| I + AI EtOH<br>(150 mg/kg)  | $0.35 \pm 0.02^{a}$ | $0.47 \pm 0.02^{a}$ | $0.52 \pm 0.04^{a}$ | $0.61 \pm 0.06^{a}$ | $0.74 \pm 0.05^{a}$ | $0.66 \pm 0.07^{a}$        | $0.66 \pm 0.06^{a}$          | $0.64 \pm 0.07^{a}$      | 39.16%                        |
| I + AI EtOH<br>(300 mg/kg)  | $0.35 \pm 0.03^{a}$ | $0.57 \pm 0.04^{a}$ | $0.61 \pm 0.01^{a}$ | $0.64 \pm 0.02^{a}$ | $0.69 \pm 0.06^{a}$ | $0.65 \pm 0.04^{a}$        | $0.64 \pm 0.04^{a}$          | $0.62 \pm 0.04^{a}$      | 42.67%                        |
| I + Ibuprofen<br>(400mg/kg) | $0.38 \pm 0.02^{a}$ | $0.47 \pm 0.02^{a}$ | $0.55 \pm 0.02^{a}$ | $0.64 \pm 0.03^{a}$ | $0.73 \pm 0.02^{a}$ | 0.65 ± 0.02 <sup>a</sup> ` | $0.62 \pm 0.02^{a}$          | 0.55 ± 0.01 <sup>ª</sup> | 64.70%                        |

Values (means of six replicates) followed by different letters are significantly different at p < 0.05

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### DISCUSSION

The present study revealed the antioxidant (*in vitro*) and anti-inflammatory activities of hexane and ethanolic stem bark extract of *Anogeissus leiocarpus* (AI) in wistar albino rats. Free radicals can react with membrane lipids, nucleic acids, proteins and enzymes, and other small molecules, causing cellular damage. To find out the antioxidant ability of hexane and ethanolic stem bark extract of *Anogeissusl leiocarpus* various antioxidant assays were carried out.

DPPH radical scavenging activity is one of the most important assessments of antioxidant potency of pure compounds or crude plant extract (Saira *et al.*, 2020). In DPPH radical scavenging assay, the hexane extract showed excellent activity, the hexane extract showed higher percentage potential ( $80.95 \pm 0.86^{\text{b}}$ ) at 100 µg/ml; IC<sub>50</sub> value was 55.17 ± 2.18 µg/ml, while ethanolic extract has lower percentage potential of  $60.26 \pm 0.18^{\text{e}}$  at 100 µg/ml and IC<sub>50</sub> value of  $80.10 \pm 6.32$  µg/ml as shown in Tables 1and 6, all in aconcentration dependent manner.

In this study, both hexane and ethanolic stem bark extract of *Anogeissus leiocarpus* (AI) quenched hydroxyl radicals and halted 2- deoxyribose breakdowns at higher concentration of 100  $\mu$ g/ml (Table 2). The higher percentage hydroxyl scavenging ability was recorded in hexane extract of stem bark of Al at 100  $\mu$ g/ml (80.70 ± 0.22<sup>c</sup>); IC<sub>50</sub> value was 47.50 ± 1.33  $\mu$ g/ml followed by hexane extract at 75  $\mu$ g/ml(70.12 ± 0.22<sup>d, e</sup>); while lower percentage hydroxyl scavenging ability was recorded by ethanolic extract of AI (61.19 ± 0.62<sup>9</sup>); IC<sub>50</sub> value was 47.94 ± 4.29  $\mu$ g/ml. It should be noted that both hexane and ethanolic extract of AI have similar IC<sub>50</sub> values as shown in Table 6. The research study showed that a concentration dependent manner for hydroxyl radical quenching activity is noticed (Table 2).

In this study, the lower value of IC<sub>50</sub> for nitric oxide scavenging ability is recorded by hexane extract of Al (56.70 ± 0.99µg/ml) with higher percentage scavenging activity at 100 µg/ml(82.64 ± 0.55<sup>c</sup>) while IC<sub>50</sub> value for ethanolic is (65.56 ± 31.92 µg/ml) at lower percentage scavenging activity at 100 µg/ml (58.92 ± 0.18<sup>h</sup>) as revealed in Tables 3 and 6, in dose dependent manner.

Both the hexane ( $48.11 \pm 6.84$ ) and ethanolic ( $46.23 \pm 5.53$ ) extracts showed appreciable IC<sub>50</sub> values that compared well with Trolox ( $41.38 \pm 6.39$ ) as shown in Table 5. At 100µg/ml, the ABTS radical scavenging capacities of hexane and ethanolic stem bark extract of *Anogeissus leiocarpus* were 85.21% and 53.37% respectively (Table 4). The extracts demonstrated a concentration dependent ABTS scavenging capacity.

Numerous studies suggested that phenolic compounds are responsible for effective free radical scavenging and antioxidant activities (Sanjay *et al.*, 2020). Therefore, determination of the phenolic content is very important in order to determine the antioxidant capacity of plant extracts (Sanjay *et al.*, 2020). In the present study, higher concentration of total phenolic content was found in hexane stem bark extract of Al (69.84  $\pm$  0.16 (mg GAE/ g extract)) while ethanolic stem bark extract recorded lower concentration of total phenolic content (53.28  $\pm$  0.15 (mg GAE/ g extract) though suggesting the importance of stem bark extract of *Anogeissus leiocarpus* for effective antioxidant activities in traditional medicine.

Also, it was already reported that naturally occurring flavonoids compounds have free radical scavenging properties, due to their hydroxyl groups, which are effective hydrogen donors, which make them antioxidant. Higher total flavonoid content ( $60.75 \pm 0.31$  (mg QUE/ g extract)) as well as total antioxidant capacity ( $70.00 \pm 2.15$  (mg AAE/g extract) were also recorded by in hexane stem bark extract of Al while ethanolic stem bark extract recorded lower concentrations of total flavonoid content ( $50.34 \pm 0.42$ mg QUE/ g extract)) and total antioxidant capacity ( $58.70 \pm 0.31$  (mg AAE/ g extract).

The Ferric reducing-antioxidant property(FRAP) values of ethanolic stem bark extract (46.67  $\pm$  0.33 (mg AA/g extract)) was lower than the hexane stem bark extract (80.55  $\pm$  1.97 (mg AA/g extract) of *Anogeissus leiocarpus* as seen in Table 6.This is indicative that the stem bark extracts of the studied plant act as reductone that reduced the ferric tripyridyltriazine [Fe (III)-TPTZ] complex to ferrous tripyridyltriazine [Fe (II)-TPTZ]. This was similar to the previous study reported which revealed that, the higher the FRAP value the higher the reducing power of the extract on electron transfer ability towards the FRAP reagents (Amaeze *et al.*, 2011).

The toxicity study which is fundamental for an adaptation of the traditional medicine was conducted to identify the tolerance limit of the study plant. According to Lorke's method, the studied plant is safe and nontoxic to rats. ( $LD_{50}$ > 5000 mg/kg).Daily oral administration of varying doses

(75,150,300 mg/kg bw) of hexane and ethanolic stem bark extract of *Anogeissus leoicarpus* was done for 14 days.No mortality was recorded after fourteen (14) days of treatment. The control and extract treated groups of animals were normal and did not show any symptoms associated with toxicity. Kabare *et al.* (2010) reported aqueous decoction of *A. leiocarpus* leaves to be between 500 to 5000mg /kg body weight. This report confirmed the low toxicity obtained in the study of hexane and ethanolic stem bark of *A. leiocarpus*.

Formalin-induced experimental models produce a distinct biphasic nociception in the paw, a first phase (lasting the first 5 mins) corresponding to acute neurogenic pain, and a second phase (lasting from 15 to 30 mins) corresponding to inflammatory pain responses as stated by Huang *et al.* (2011). In the present study, notable oedema and inflammation was observed in the experimental animals. Oedema formation in the paw is as result of a synergism between various inflammatory mediators that increased vascular permeability and blood flow as reported by John and Shobana (2012). However, pretreatment and daily oral administration with hexane and ethanolic stem bark extract *Anogeissus leiocarpus*. showed a significant inhibition in the late phase of formalin induced pain in a dose dependent manner which was well comparable with the standard drug ibuprofen and also corroborates with the findings of Young *et al.* (2005).

The results of this study revealed that stem bark extract of as well as the standard drug (ibuprofen) significantly reduced the rat right hind paw oedema induced by subcutaneous injection of formalin, suggesting marked anti-inflammatory potentials of the studied plant in dose-dependent manner.

In conclusion, the presence of phytochemicals in the study plant can explain the antiinflammatory activity of this study plant. Hexane and ethanolic stem bark extract of *Anogeissus leiocarpus* have revealed the significant antioxidant and anti-inflammatory potentials, thereby establishing the pharmacological basis for its use in the treatment of arthritis, inflammation, oedema and some diseases associated with oxidative stress in Nigeria ethno medicine.

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# REFERENCES

Ashmawi HA, Freire GMG.(2016). Peripheral and central sensitization. RevistaDor, 17. (1): 1-4.

- Asahara M, Ito N, Yokomizo T, Nakamura M, Shimizu T, Yamada Y (2015). The absence of the leukotriene B4 receptor BLT1 attenuates peripheral inflammation and spinal nociceptive processing following intraplantar formalin injury. Molecular Pain11:1-12.
- Chen L, Deng H, Cui H, Fang J, Zuo Z, Deng J, Li Y, Wang X, Zhao, L (2017). Inflammatory responses and inflammation-associated diseases in organs. Oncotarget, 9(6): 1-15.
- Dzeufiet DPD, Nguelefack TBA, Jacqueline C, Gérard KP,Dimo T (2014). Anti-Diabetic activity of Terminalia superba( Combretaceae) Stem Bark Extract in Streptozotocin Induced Diabetic Rats. British Journal of Pharmaceutical Research 4(11): 1300-1310.
- Huang MH, Huang SS, Wang BS, Wu CH, Sheu MJ, Hou WC (2011). Antioxidant and antiinflammatory properties of Cardio spermum halicacabum and its reference compounds ex vivo and in vivo.J Ethnopharmacol. 133(2):743-750.
- Hung Y, Sava V, Yang SM, Hong MY, Yang PC, Huang GC (2001). Isolation and characterization of melanin pigments derived from tea and tea polyphenols. Food Chem. 73(2): 177-184.
- Jang Y, Kim, M Hwang SW (2020). Molecular mechanisms underlying the actions of arachidonic acid-derived prostaglandins on peripheral nociception. Journal of Neuroinflammation. 17(30) :1-27.
- Ji RR, Nackley A, Huh Y, Terrando N, Maixner W (2018). Neuroinflammation and Central Sensitization in Chronic and Wide spread Pain. Anesthesiology, 129 (2), 343–366.
- Jin M, Cai Y, Li J, Zhao H (1996). 1, 10-Phenanthroline-Fe2+ oxidative assay of hydroxyl radical produced by H2O2/Fe2+. Prog. Biochem. Biophys. 1996: 23, 553–555.
- Joseph SM, George MC, Nair JR (2005). Effect of feeding cuttlefish liver oil on immune function,

inflammatory response and platelet aggregation in rats. Current Sci; 88(3):507-10.

- John NAA, Shobana G (2012). Anti-inflammatory activity of *Talium fructicosum* L. on formalin induced paw oedema in albino rats. Journal of Applied Pharmaceutical Science. 2(1): 123-127.
- Kumaran A, Karunakaran RJ (2007). *In vitro* antioxidant activities of methanol extracts of five Phyllanthus species from India. LWT Food Sci. Technol. 40: 344–352.
- Lorke D (1983). A New Approach to Practical Acute Toxicity Testing. Achieves of Toxicology 54: 275-286.
- Molyneux P (2004). The use of the stable free radical diphenylpicryl-hydrazyl (DPPH) for estimating antioxidant activity. J. Sci. Technol. 26: 211-219.
- Organization of Economic Cooperation and Development (OECD) (2001). Guidelines for testing of chemicals. Revised draft 425, Document on acute oral toxicity class method, revised Dec., 2001.
- Puntel RL, Nogueira CW, Rocha JB (2005). N-Methyl-Daspartate receptors are involved in the quinolinic acid, but not in the malonate pro-oxidative activity *in vitro*. Neurochemical Research. 30(3):417–424.
- Re R, Pellegrin N, Proteggente A, Pannala A, Yang M, Rice-Evans C, (1999). Antioxidant activity applying an improve ABTS Radication decolourization assay. Free Rad. Biol. Med. 26: 1231-1237.
- Saira B, Munazza A, Huma FH Muhammad RK (2020). Evaluation of antioxidant and antiinflammatory potency of *Lepidium pinnatifidum Ledeb*. Journal of Clinical Phytoscience. 6(21):1-12.
- Sangameswaran B, Balakrishnan BR, Deshraj C, Jayakar B (2009). *In vitro* Antioxidant activity of Thespesia Lampas Dalz and Gibs.Pakistan Journal of Pharmacy Science. 22: 368 372.
- Taylor BK, Peterson MA, Roderick RE, Tate J, Green PG, Levine JO, Basbaum AI (2000). Opioid inhibition of formalin-induced changes in plasma extravasation and local blood flow in rats. Pain 84: 263–270.